

Optically-Based Diagnostics for Gas-Phase Laser Development

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Overview



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- Theme: multi-species diagnostics for absolute concentrations are essential for effective development of high-energy gas lasers
 - Precursor production, loss, optimization
 - Transference from subscale reactors
 - Scaling of gain, power, efficiency

Current applications

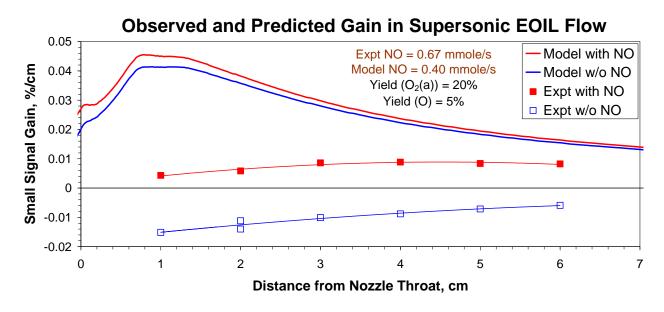
- Electric Oxygen Iodine Laser (EOIL): precursor kinetics and gain dynamics
- Related systems: COIL, micro-COIL
- Alkali laser systems: DPAL, XPAL gain, multi-photon effects

Outline of presentation:

- Brief overview of diagnostics and apparatus
- Absolute emission spectrometry
 - Near-infrared spectroscopy: O₂(a¹Δ_q), I(²P_{1/2})
 - Air afterglow photometry: O(3P)
- Ultrasensitive absorption photometry: I₂, O₃
- High-resolution absorption/gain spectroscopy: atomic iodine, alkali metals

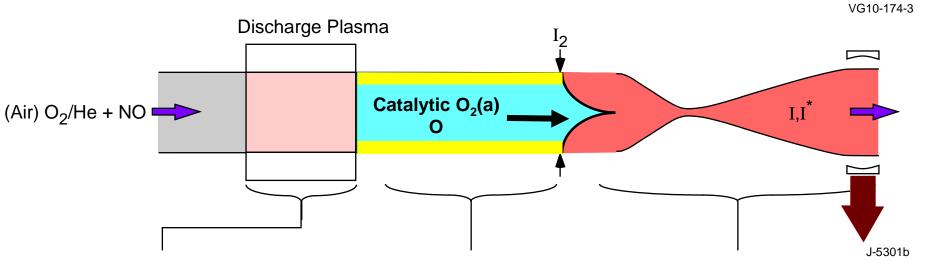
Role of Optical Diagnostics in High Energy Gas Laser Development

- Chemically rich, energetic, reacting flow with competing phenomena
 - Multispecies detection required
 - COIL, EOIL: O₂(a,b), I*, I, I₂, gain, T, O, O₃
 - DPAL, XPAL: ground-state M, numerous M*, M₂*, MX*, gain
- Objective: detect key species concentrations vs. flow time
 - Vary operating conditions systematically
 - Quantify species production and loss rates
 - Relate to system design requirements
- Put a "cage" around the model:



Electric Oxygen Iodine Laser (EOIL)





$$e^{-} + O_{2} \rightarrow O_{2}(a) + e^{-}$$

$$e^- + O_2 \rightarrow O + O + e^-$$

Ionization

$$e^- + O_2(a) \rightarrow losses$$

Dilution in He required

$$O + NO + M \rightarrow NO_2 + M$$

$$O + NO_2 \rightarrow NO + O_2$$

$$O + O_2 + M \rightarrow O_3 + M$$

$$O+O_3 \rightarrow O_2+O_2$$

$$NO + O_3 \rightarrow O_2 + NO_2$$

$$O+I_2 \rightarrow IO+I$$

$$O + IO \rightarrow O_2 + I$$

$$O_2(a) + I \rightleftharpoons O_2 + I^*$$

$$O + I^* \rightarrow O + I$$

$$I^* + O_2 \rightarrow I + O_2(v)$$

$$\frac{\left[I^{*}\right]}{\left[I\right]} \rightarrow K_{EQ}(T) \frac{\left[O_{2}(a)\right]}{\left[O_{2}\right]}$$

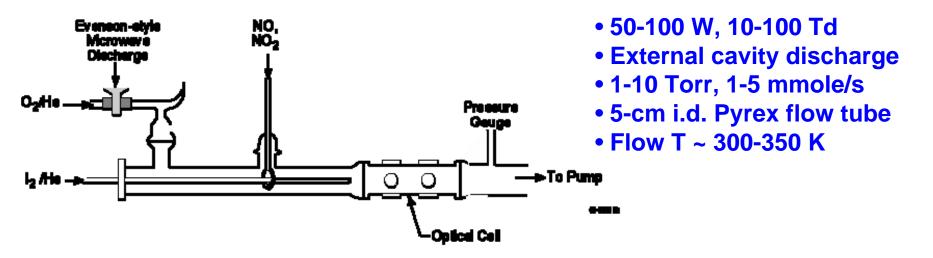
Hybrid EOIL: Catalytically enhanced O₂(a)

PSI Microwave Discharge Flow Reactors (2450 MHz)

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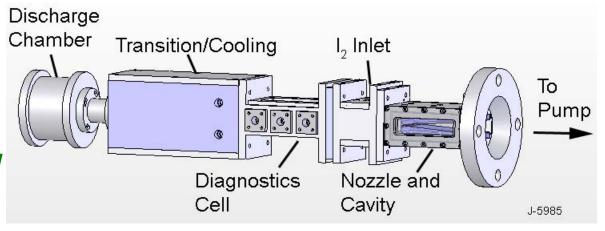
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Low-Pressure Reactor: Active-O₂ Kinetics



EOIL Subsonic/Supersonic Reactor

- 1-5 kW, 10-50 Td
- Coaxial MIDJet discharge
- 30-70 Torr, 40-100 mmole/s
- M ~ 2 supersonic cavity
- Lasing typically 100-150 mW
 - $M^2 = 1.08 \pm 0.01$

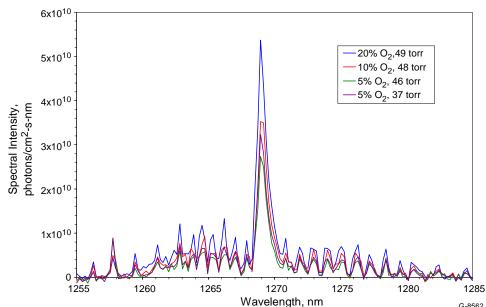


Near-IR Absolute Emission: O₂(a) and I*

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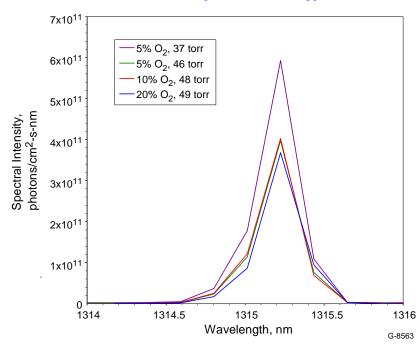
- InGaAs array monochromator: observe entire band
- Concentration = Intensity ÷ Einstein coefficient

$$O_2(a^1\Delta_g \rightarrow X^3\Sigma_g^-)$$



- $A_{00} = 2.20 \times 10^{-4} \text{ s}^{-1} (\pm 10\%)$
- Detection limit ~ 5 x 10¹² cm⁻³
 (5 cm path)

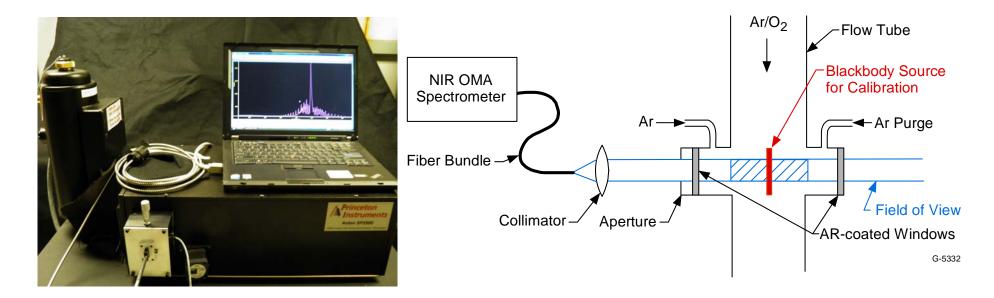
$$I(^{2}P_{1/2} \rightarrow ^{2}P_{3/2})$$



- $A = 8.0 \text{ s}^{-1} (\pm 20\%)$
- Detection limit ~ 10⁸ cm⁻³
 (5 cm path)

NIR Photometric Calibration: O₂(a) and I* Emission

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- Collimated field of view: no reflective surfaces
 - Eliminate stray light, e.g. discharge emission
- Etendue (AΩ) for calibration is identical to that for volume emission
- Spectral responsivity = (signal) / (Planck function)
- Blackbody calibrations 800-1000° C agree within 1%

Principle of Absolute Calibration Method

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Instrumental signal for both blackbody and $O_2(a \rightarrow X)$ emission:

• $S(\lambda) = F(\lambda) T(\lambda) A\Omega \delta\lambda I(\lambda)$ F = spectral responsivity

• Use the same $\{A\Omega \delta \lambda\}$ for calibration and gas emission measurements

Determine F by measuring blackbody spectrum (Planck function):

• $F(\lambda) = S_{BB}(\lambda) / N(\lambda, T_{BB})$ $T_{BB} \sim 1000 C$

- Area of source > area of fov
- Accuracy <1%

Gas radiance in photons/cm²-s-sr-nm:

- $I_{ax}(\lambda) = S_{ax}(\lambda) / \{F(\lambda) T(\lambda)\}$
- Correct for spectral baseline/background

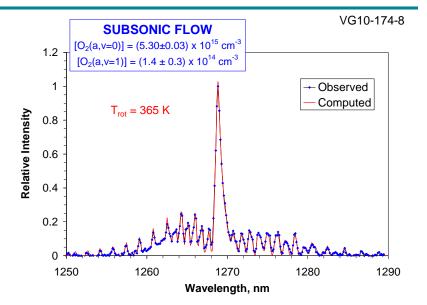
Determine $O_2(a)$ concentration from spectrally integrated intensity:

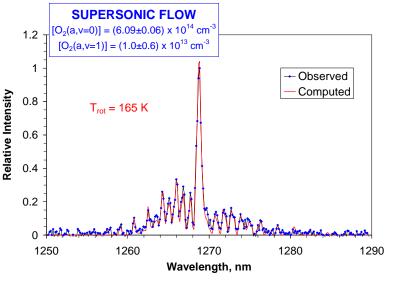
•
$$[O_2(a)] = (4\pi/\ell) \int I_{aX}(\lambda) d\lambda / A_{aX}$$
 $A_{aX} = Einstein coefficient$

Spectral Fitting Analysis: [O₂(a,v)], T_{rot}

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- $O_2(a^1\Delta_g \rightarrow X^3\Sigma_g^-)$ spectroscopy:
 - Magnetic dipole transition
 - Hund's coupling case (b)
 - Bose-Einstein statistics (¹6O₂)
 - → 9 rotational branches
- Our procedure: determine line strengths for (0,0) from line-by-line compilation (HITRAN)
 - Boltzmann rotational temperature
 - Shift (0,0) envelope to band centers for (1,1), (2,2), etc.
- Convolve with instrument scan function
 - Triangular slit function for grating monochromator (0.3 nm FWHM)
- Linear least squares solutions are {[O₂(a,v]}





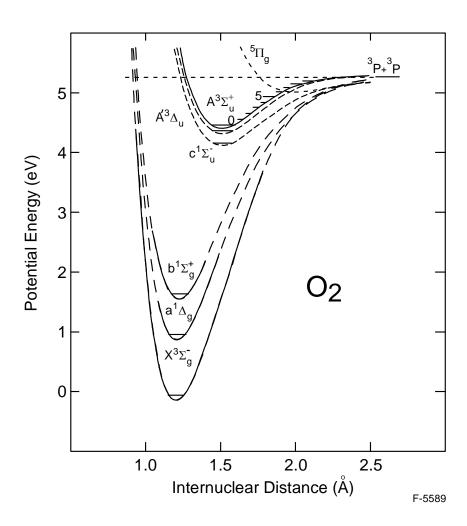
Estimation of A₁₁, A₂₂, A₃₃ Values



 Scale values from A₀₀ via Franck-Condon factors, transition moment vs. r-centroid:

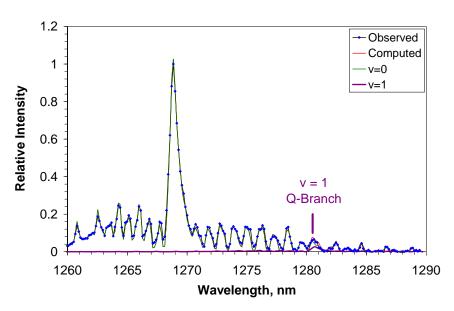
$$A_{v'v''} = (64\pi^4/3h) (v^3q_{v'v''}) (R(r))^2$$

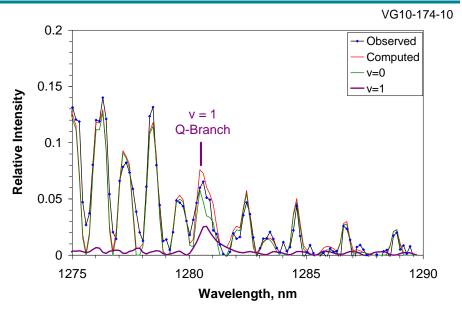
- Franck-Condon factors, r-centroid values from Krupenie (1972)
- Estimate scaling of (R(r))² from A₀₀:A₀₁:A₁₀
 - Literature: A_{00}/A_{01} is either ~50 or ~80
 - PSI measurement: $A_{00}/A_{01} = 52 \pm 6$
 - Badger et al. (1965): $A_{00}/A_{10} > 200$
 - Solution: (R(r))² varies slightly with v'
- Solutions for $A_{00} = 2.20 \times 10^{-4} \text{ s}^{-1}$: $A_{11} = 2.17 \times 10^{-4} \text{ s}^{-1}$ $A_{22} = 2.12 \times 10^{-4} \text{ s}^{-1}$ $A_{33} = 2.06 \times 10^{-4} \text{ s}^{-1}$



BUT WE DO NOT OBSERVE $O_2(a,v>0)$!



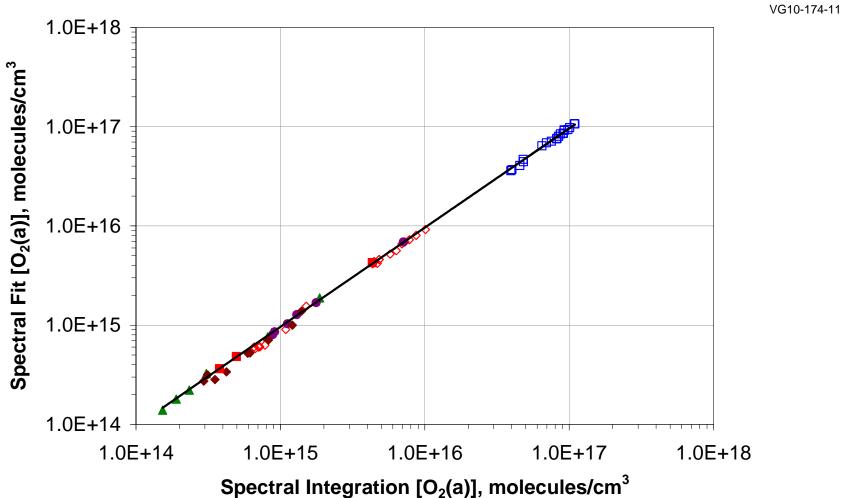




- Typical fits: [(v=1)]/[(v=0)] ~ 3 to 7 %
 - Tends to track with temperature of discharge, i.e. thermal populations only
- True for large range of conditions:
 - 50 W 2 kW discharges
 - 0.5 50 Torr
 - Cl₂/BHP generators, energy pooling conditions
- Slanger, Copeland 2003: O₂(a,v) exchange with O₂(X, v=0) is fast
- Implications for COIL I₂ dissociation mechanism?

Spectral Fitting vs. Integration





Philosophy: spectral fitting confirms band shape, T_{rot}, T_{vib}, no other radiators; then integration gives accurate values

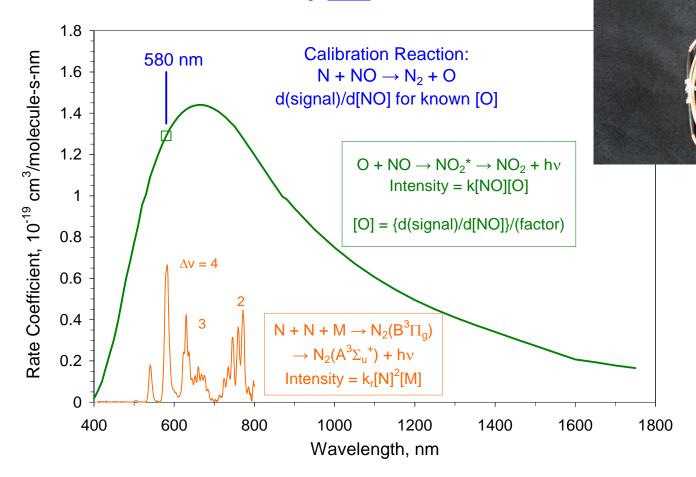
Air Afterglow: $O + NO \rightarrow NO_2 + hv$

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(e)

- Fiber-coupled photomultiplier
 -- 580 nm filter, collimated field-of-view
- Calibrate with blackbody <u>and</u> titration reaction

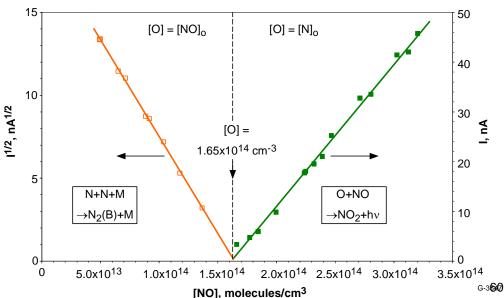


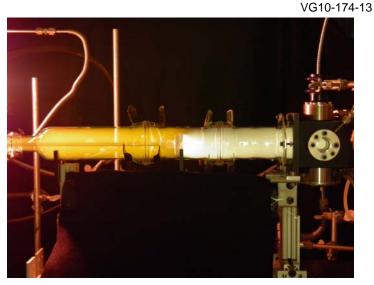
J. Phys. Chem. 90, 320-325 (1986)

- Calibrations allow measurement of absolute emission rates for known [O], [NO]
- → Determination of k(580 nm)
- Scale to other wavelengths via relative intensity measurements

Air Afterglow: Determination of [O]

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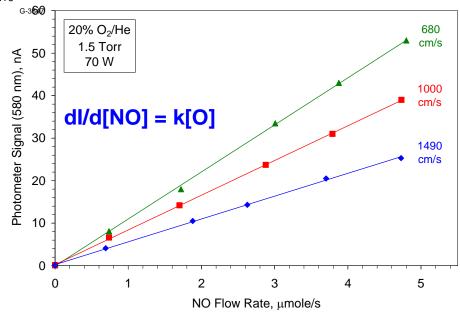




 N_2 Discharge: $N + NO \rightarrow N_2 + O$ $O + NO \rightarrow NO_2 + hv$

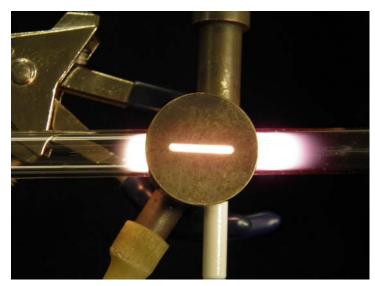
For [NO] < [N]: N₂(B→A) emission For [NO] > [N]: O + NO emission slope ÷ [O] = calibration factor correct for O+NO+M reaction

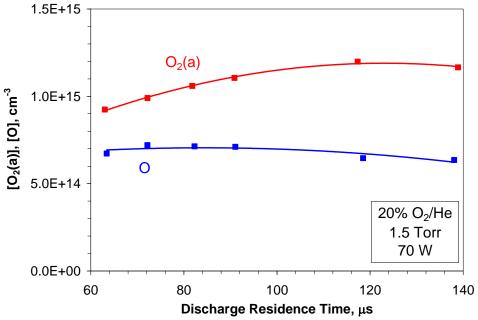
Blackbody calibration: $\rightarrow k_{580}$

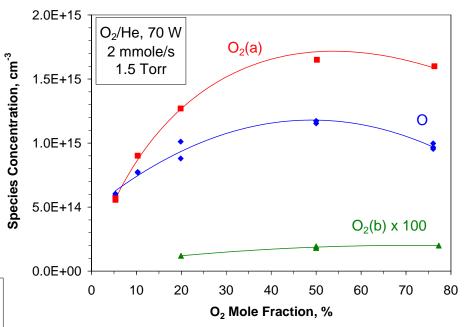


Measurements: Discharge Production of O, O₂(a)









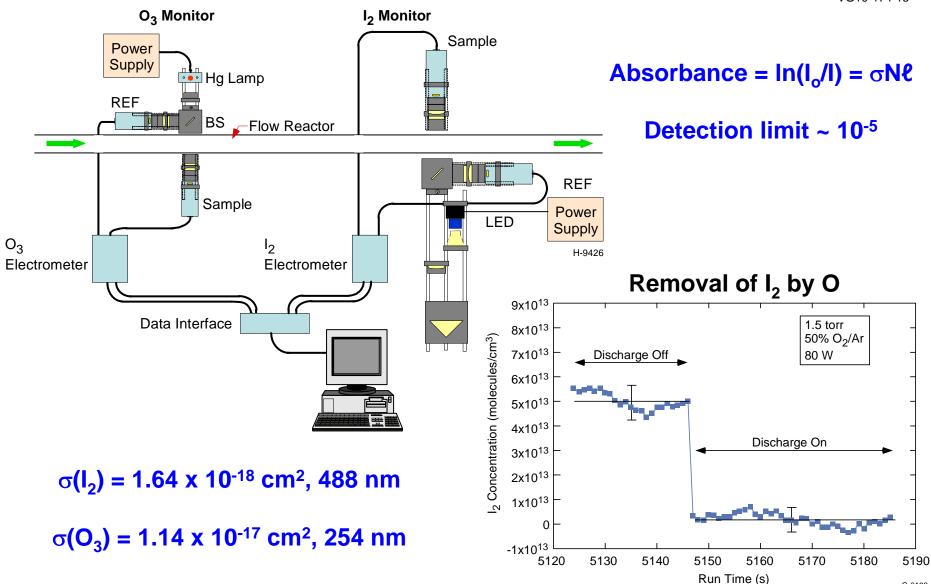
- Observed [O]/[O₂(a)] is ~ 1 or less
- Discharge models predict[O] >> [O₂(a)]
 - Electron-impact O₂ dissociation cross sections are too large
 - Possible O loss on hot walls

Ultrasensitive Dual-Beam Absorption: O₃ and I₂

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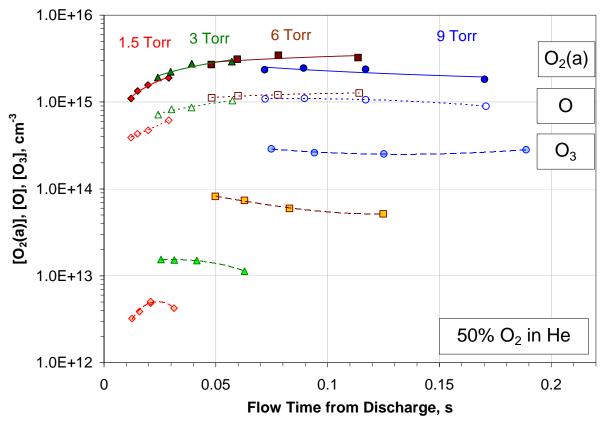
G-0183



O₃ Formation in Active-Oxygen Flow



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Surprise: O and O₃ are in ~ steady state!
Requires O₃ conversion to O

$$O + O_2 + M \rightarrow O_3(v) + M$$

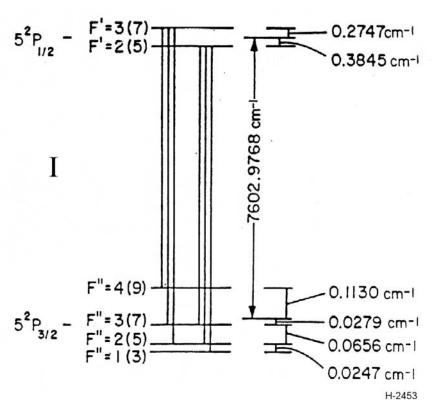
 $O_2(a) + O_3(v) \rightarrow O + 2O_2$, $k = 5 \times 10^{-11} \text{ cm}^3/\text{s}$

J. Chem. Phys. <u>87</u>, 5209-5221 (1987)

Small Signal Gain: Atomic Iodine

BIP

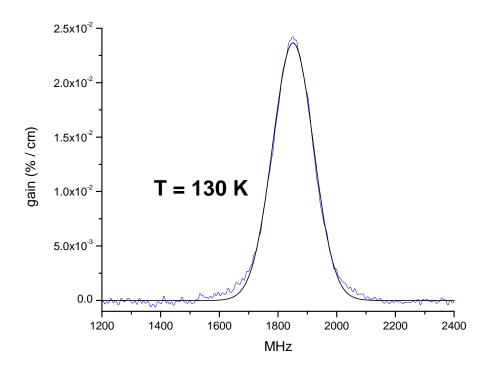
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- Probe transmission on (3,4) line
- $G/\sigma(T) = [I^*] [I]/2$
- [I*] from IR emission

 → [I], [I*]/[I], ([I*]+[I])/2

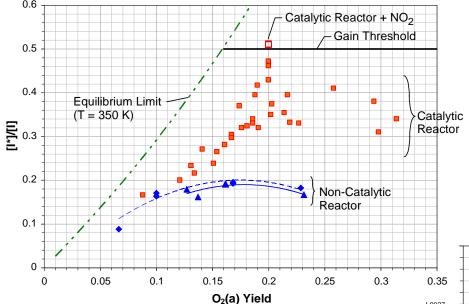
- Scanning tunable diode laser
- Balanced ratiometric detection
- Detection limit ~10⁻⁵ %/cm
- Doppler width → temperature
- Method widely used for COIL, EOIL development



Observations of I*/I Behavior







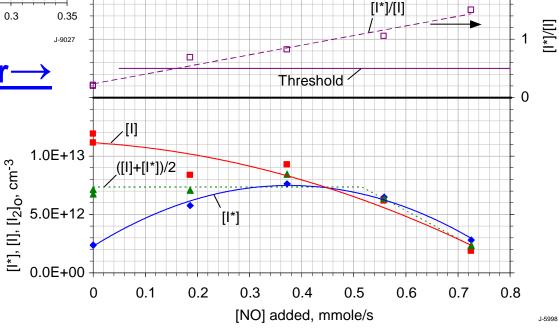
←Subsonic flow reactor

- I*/I ratio is limited by I* loss reaction
- Catalytic environment enhances attainable I*/I
- Chem. Phys. Lett. <u>469</u>, 68-70 (2009)
 Proc. SPIE 7196-04 (2009)

Proc. SPIE 7581-06 (2010)

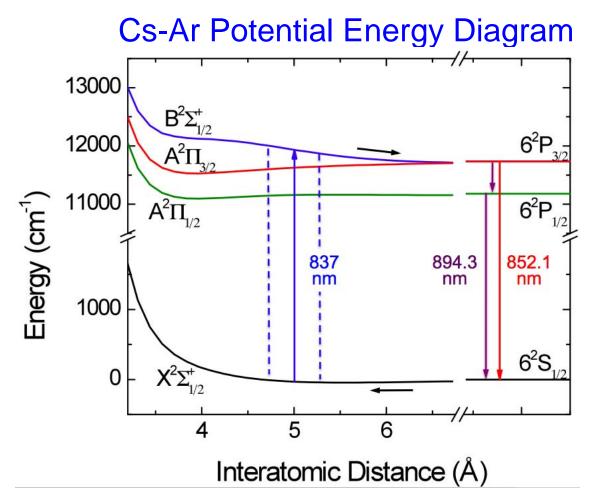


- I*/I ratio is limited by I* loss reaction
- Addition of NO enhances I*/I
- I*/I continues to increase past optimum gain point
- Proc. SPIE 6874-10 (2008)
 Proc. SPIE 7581-03 (2010)
 J. Appl. Phys. D 43 025208 (2010)





Application to Alkali and Alkali-Exciplex Systems



- DPAL: pump D₂, lase D₁ (C₂H₆ promotes spin-orbit transfer)
- XPAL: pump broadband exciplex X→B, lase on D₁ or D₂

DPAL/XPAL Gain Measurement Test Bed (Diode laser scanning D₁ line)

Ti: S Laser

Alkali Cell

Filter

Signal

Detector

Reference

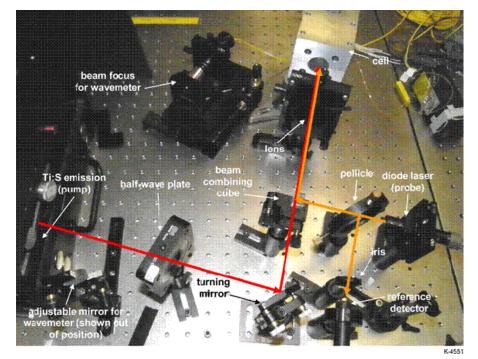
Detector

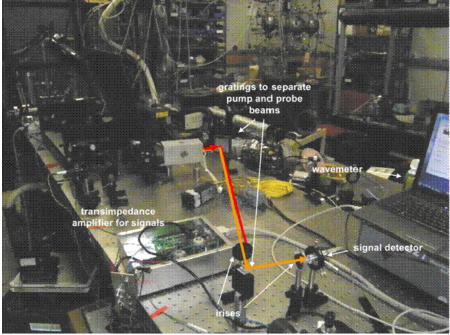
Scope

K-1858a

- Direct probe of population inversion dynamics
- Aids in design of optical resonators
- Portable: take to other facilities
- Can extend to spatial imaging of gain
 - Expect significant spatial effects in power scaling
 - Valuable tool for scaling DPAL to high powers

Optical Layout for DPAL/XPAL Gain Measurements Physical Sciences Inc.





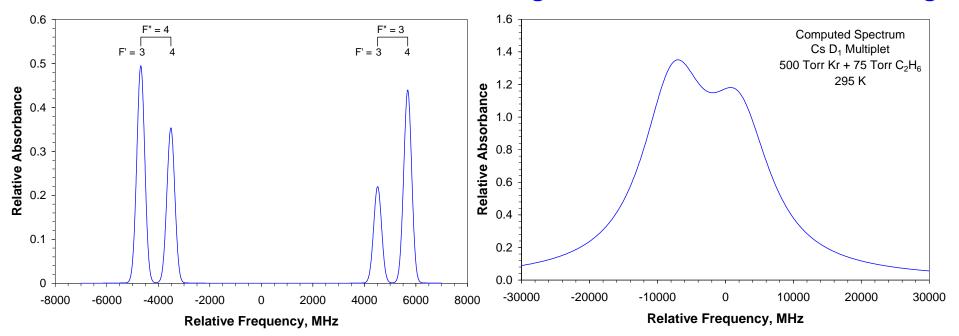
Computed D₁ Absorption Spectra: Cs Collisional Broadening Effect Physical Sciences Inc.

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$$Cs {}^{2}S_{1/2} - {}^{2}P_{1/2}$$
, 894 nm

Low Pressure, Doppler broadening

High Pressure, collisional broadening

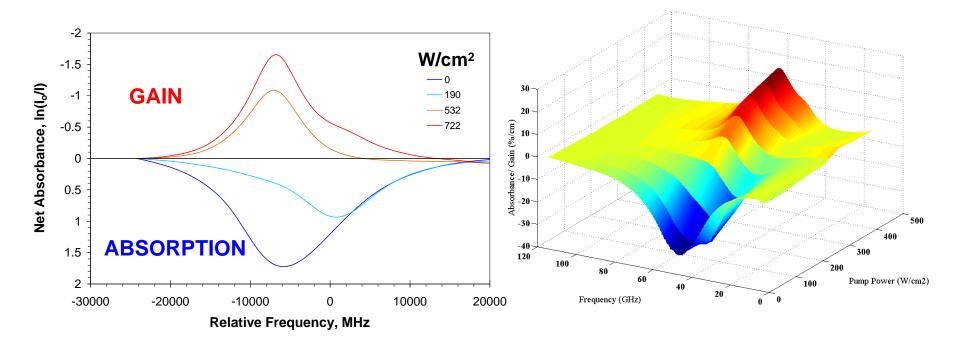


- Collisional broadening greatly expands required scan range
- High optical thickness at elevated temperatures

Absorption/Gain Spectra: $Cs(^2S_{1/2},F''=4\rightarrow^2P_{1/2},F')$, 894 nm 500 Torr Kr + 75 Torr C_2H_6 , 338 K

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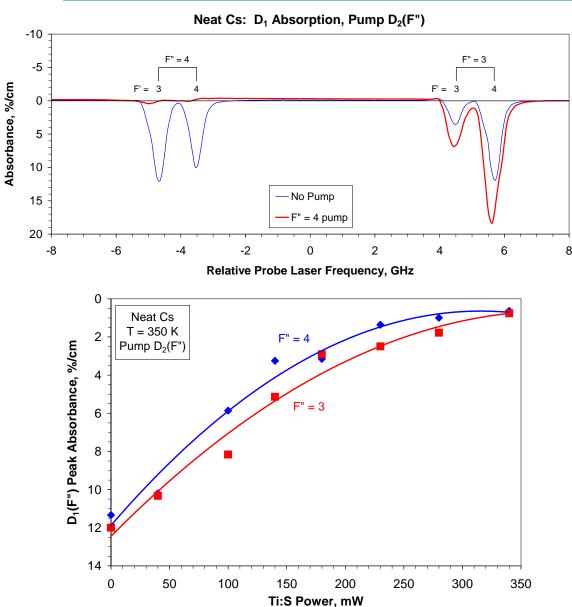
Pump Laser: ${}^{2}S_{1/2} \rightarrow {}^{2}P_{3/2}$, 852 nm



 Continuing work: investigate absorption and gain dynamics for DPAL, XPAL configurations: Cs, Rb, K

State-Selected Absorption and Saturation

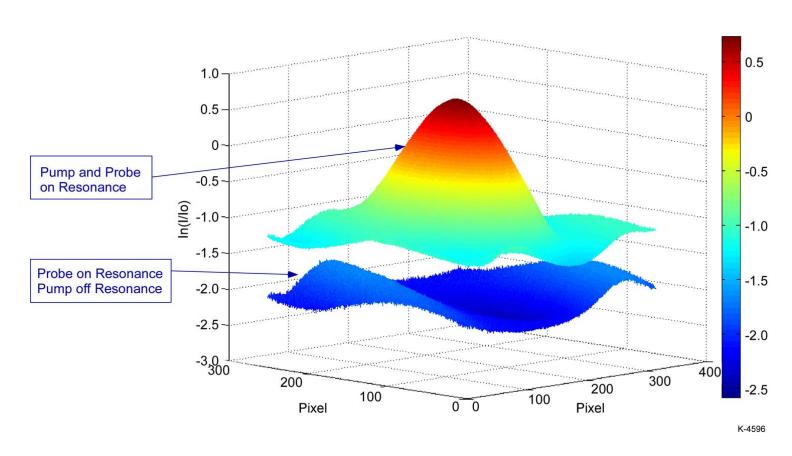




3-D Image of D₁ Gain, Absorption Cs + 500 Torr Kr + 75 Torr C₂H₆

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- Probe beam diameter > pump beam diameter
- Sample 9 combinations: probe{on peak, off peak, blocked} x pump{on peak, off peak, blocked}



Gain profile follows Gaussian profile of pump beam

Conclusions



- Multispecies diagnostic suite
 - Absolute emission spectrometry
 - Ultrasensitive absorption photometry
 - Scanning TDL absorption/gain spectroscopy
- Extreme sensitivity enables subscale operation at low species concentrations
 - Simplify chemistry, focus on primary reaction steps
 - Transfer to large scale systems: establish models for scaling
- EOIL, catalytic EOIL, COIL, micro-COIL: operational parametrics
 - O₂(a) yield vs. small-signal gain
 - I_2 dissociation: $[I_2]$, $[I^*]$ + [I]
 - O, O₃ effects
- DPAL, XPAL: power scaling phenomena
 - Gain vs. pump power, spatial effects at high optical depth
 - General emission spectroscopy: multi-photon effects vs. pump power

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